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## 5 The Periodic Law

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Chapter 5 - The Periodic Law 5-1 History of the Periodic Table I. Mendeleev's Periodic Table A. Organization 1. Vertical columns in atomic weight order a. Mendeleev made some exceptions to place elements in rows with similar properties (1) Tellurium and

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iodine's places were switched 2. Horizontal rows have similar chemical properties

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CHAPTER 10 REVIEW States of Matter SECTION 1 SHORT  
ANSWER Answer the following questions in the space provided.  
1. Identify whether the descriptions below describe an ideal gas  
or a real gas. ideal gas a. ... SECTION 5 MODERN CHEMISTRY  
STATES OF MATTER. STATES OF MATTER MODERN CHEMISTRY.

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CHAPTER 5 REVIEW The Periodic Law SECTION 1 SHORT ANSWER  
Answer the following questions in the space provided. 1. c In the modern periodic table, elements are ordered (a) according to decreasing atomic mass. (b) according to Mendeleev's original design. (c) according to increasing atomic number. (d) based on

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when they were discovered.

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## Study GuideChapter 5-21 Answer Key

Reviewed by Andrew Bonham, Professor, Metropolitan State University of Denver on 7/16/19 Modern chemistry chapter 6 5 review answers. This textbook addresses all the major topics that I would expect to see in a General, Organic, and biological Chemistry textbook.

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The following pages contain the bulk (but not all) of the information for the chapter 5 test. Focus on this content, but



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make sure to review class notes, activities, handouts, questions, etc. If you study this document and NOTHING else, you should at least be able to PASS the test.

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CHAPTER 6 REVIEW Chemical Bonding SECTION 5 SHORT

ANSWER Answer the following questions in the space provided.

1. Identify the major assumption of the VSEPR theory, which is used to predict the shape of atoms. Pairs of valence electrons repel one another. 2. In water, two hydrogen atoms are bonded to one oxygen atom. Why isn't water a linear ...

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