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Explore the relationship between molecular motion, temperature, and phase changes. Compare the molecular

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structure of solids, liquids, and gases. Graph temperature changes as ice is melted and water is boiled. Find the effect of altitude on phase changes. The starting temperature, ice volume, altitude, and rate of heating or cooling can be adjusted.

Student Exploration: Phases of Water

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as ice is melted and water is boiled. Find the effect of altitude on phase changes.

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Questions & Answers 1. The graph
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Micro view and set the Ice volume to 50 cc. Click Play and observe molecules in the solid/ice, liquid/water, and gas/air phases. Record your observations in your science notebook. Here are some example questions to get you started: In which phase are the molecules held tightly together? In which phase are the molecules moving freely?

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Get the Gizmo ready: Click Reset, and select the Micro view. Set Ice volume to 0 cc. Set Add/remove heat energy to 0 J/s. Question: Why do phase changes occur? 1. Compare: Set the Water temperature to 0 °C and click Play. Observe the water molecules. Click Reset, set the Water temperature to 100

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°C, and click Play again. What do you notice?

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Student Exploration: Phase Changes
Vocabulary: altitude, boil, boiling point, freeze, freezing point, gas, liquid, melt,

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melting point, phase, solid Prior
Knowledge Questions (Do these BEFORE
using the Gizmo.) 1. A family from
Minnesota turns off the heat and flies to
Florida for a winter holiday. When

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3. A phase change is a change from one

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phase to another. What phase changes would you see in this example? Melting and boiling will be seen in this example. Condensation might be seen as well.

Gizmo Warm-up In the Phases of Water Gizmo™, you can heat up or cool down a container of water. 1. Press Heat to heat up the water. Wait until the ...

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